## Regioselective Photo-oxygenation of 2-Aryl-4,5,6,7-tetrahydroindoles: Preparation and Reactions of 2-Aryl-7ahydroperoxy-4,5,6,7,7a-pentahydroindolenines

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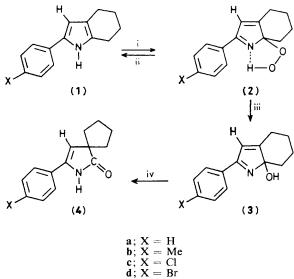
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Dye-sensitized photo-oxygenation of 2-aryl-4,5,6,7-tetrahydroindoles quantitatively afforded 2-aryl-7a-hydroperoxy-4,5,6,7,7a-pentahydroindolenines, which were converted into the corresponding hydroxy-indolenines and spiro- $\gamma$ -lactams in excellent yields.

Dye-sensitized photo-oxygenation of pyrroles has been widely investigated during the last 2 decades.<sup>1-7</sup> However, in this photoreaction, a variety of oxidation products or complex mixtures were generally produced in a non-selective manner except in the cases of poly-phenyl- and poly-t-butyl-substituted pyrroles.<sup>6,7</sup> We now report the regioselective photooxygenation of bicyclic pyrroles such as 2-aryl-4,5,6,7-tetrahydroindoles (**1a**-**d**) to give stable hydroperoxy compounds (**2a**-**d**) at room temperature.

Irradiation of methanolic solutions of (1a-d) containing methylene blue as sensitizer under a continuous stream of oxygen gas with a 300 W halogen lamp for 20—30 min regioselectively gave 2-aryl-7a-hydroperoxy-4,5,6,7,7a-pentahydroindolenines (**2a**—**d**) in quantitative yield (Scheme 1).† No solvent-incorporated product was detected *via* its endoperoxide even at low temperature ( $\leq -50$  °C).

<sup>&</sup>lt;sup>†</sup> After evaporation of methanol below 30  $^{\circ}$ C, the residue was dissolved in benzene-hexane(1:1). Insoluble methylene blue was filtered off and evaporation of the solvent below 30  $^{\circ}$ C gave the hydroperoxide as a colourless solid in a practically pure state.





Scheme 1. i, hv-methylene blue, O<sub>2</sub>, MeOH; ii, NaBH<sub>4</sub>, MeOH; iii, PPh<sub>3</sub> or Me<sub>2</sub>S; iv, heat or KOBu<sup>t</sup>-DMF.

For the formation of (2a), tetraphenylporphyrin was also effective as a sensitizer. However, a complex mixture was obtained when rose bengal or  $Ru(bpy)_3^{2+}$  (bpy = bipyridine) was used. It was also found that MeCN, CH<sub>2</sub>Cl<sub>2</sub>, and C<sub>6</sub>H<sub>6</sub> can be used as solvents. The oxidation of (1a) by NaOCl- $H_2O_2$ in an aqueous methanolic solution afforded the hydroperoxide (2a) in the dark. The formation of (2a) in the methylene bluesensitized photoreaction was quenched by 1,4-diazabicyclo-[2.2.2] octane or  $\beta$ -carotene efficiently. These results clearly suggest that  ${}^{1}O_{2}$  is an active species for the photo-oxidation.

The structures of (2a-d) were assigned from their ele-

mental analyses and spectral properties, ‡ and also from chemical transformations including the positive peroxide test.

The hydroperoxides (2a-d) were stable at room temperature ( $\leq 35$  °C) for at least 10 days. The reduction of (2a-d) with PPh<sub>3</sub> in Et<sub>2</sub>O or Me<sub>2</sub>S in MeOH gave 2-aryl-7a-hydroxy-4,5,6,7,7a-pentahydroindolenines (3a-d) in quantitative yields.<sup>‡</sup> In contrast, the reduction of (2a) with NaBH<sub>4</sub> in MeOH afforded the starting pyrrole (1a) in a quantitative yield. The other unusual observation was that the spiro- $\gamma$ lactams (4a-d) are obtained in excellent yields (80-90%) via 1,5-sigmatropic rearrangement on pyrolysis of (3a-d) at 180 °C for 3 min under an argon atmosphere or on treatment with KOBu<sup>t</sup> in N,N-dimethylformamide at -30 °C for 1 h.‡

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<sup>‡</sup> Compound (2a): m.p. 126–128 °C (decomp.); u.v.  $\lambda_{max}$ (MeOH) 252 nm ( $\epsilon$  14 000); i.r.  $\nu_{max}$ (KBr) 3040, 2840–2740 (OOH···N), 1640 (C=N), and 1600 cm<sup>-1</sup>; m/z 213 ( $M^{-16}$ ); <sup>1</sup>H n.m.r.  $\delta$ (CDCl<sub>3</sub>) 8.80–3.05(8H, m, CH<sub>2</sub>), 6.40(1H, s, CH=C), 7.00–7.60 (5H, m, Ph), and 12.00(1H, br., OOH). Compound (3a): m.p. 154–155 °C (decomp.); u.v.  $\lambda_{max}$ (MeOH) 252 nm( $\epsilon$  17 200); i.r.  $\nu_{max}$  (KBr) 3080(OH), 1635 (C=N) cm<sup>-1</sup>; m/z 213( $M^+$ ); the <sup>1</sup>H n.m.r. spectrum of (3a) was not obtained owing to its insolubility In CDCl<sub>3</sub> and (CD<sub>3</sub>)<sub>2</sub>SO. Compound (**4a**): m.p. 181–182 °C; i.r.  $v_{max}$ (KBr) 3140(NH), 1680(C=O) cm<sup>-1</sup>; m/z 213( $M^+$ ); <sup>1</sup>H n.m.r.  $\delta$ (CDCl<sub>3</sub>) 1.40–2.10(8H, m, CH<sub>2</sub>), 5.65(1H, d, J 2 Hz, CH=C), 7.20–7.60(5H, m, Ph), 9.45(1H, br, NH); <sup>13</sup>C n.m.r.  $\delta$ (CDCl<sub>3</sub>) 26.1(1), 36.5(1), 57.4(s), 111.6(d), 124.7(d), 128.6(d), 128.8(d), 130.3(s), 138.0(s), and 187.0(s) p.p.m. All new compounds gave satisfactory elemental analyses.